

Today's outline - September 30, 2024



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- Molecule scattering

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- Molecule scattering
- Liquid scattering

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- Small angle x-ray scattering

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- Calculating R_g

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Reading Assignment: Chapter 4.5; Chapter 5.1

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Homework Assignment #03:

Chapter 3: 1,3,4,6,8

due Friday, October 05, 2024

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Reading Assignment: Chapter 4.5; Chapter 5.1

Homework Assignment #03:

Chapter 3: 1,3,4,6,8

due Friday, October 05, 2024

Homework Assignment #04:

Chapter 4: 2,4,6,7,10

due Monday, October 14, 2024

Scattering from molecules



From the atomic form factor, we would like to abstract to the next level of complexity, a molecule (we will leave crystals for Chapter 5).

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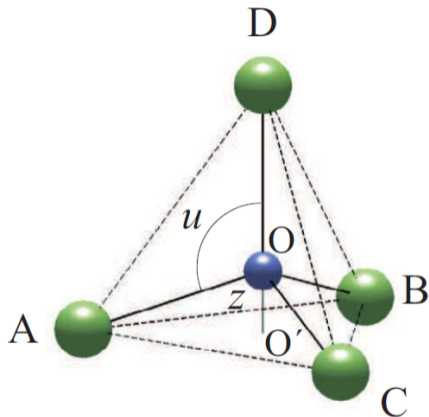
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Scattering from molecules

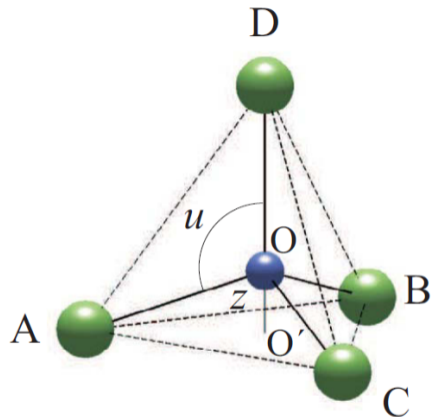


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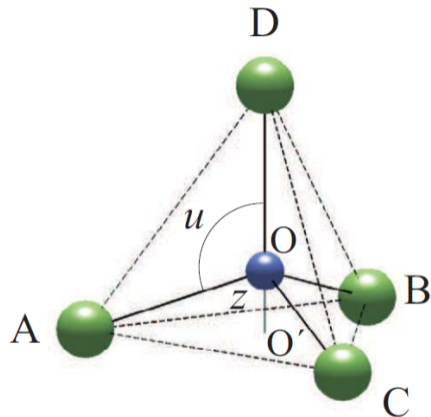
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Scattering from molecules



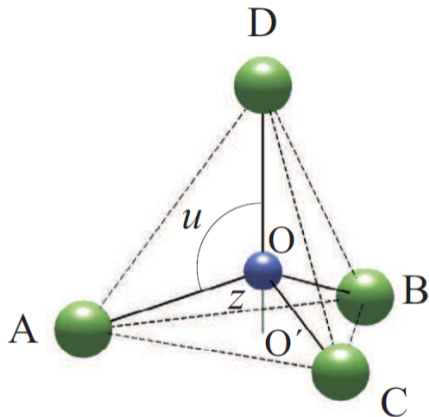
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Scattering from molecules



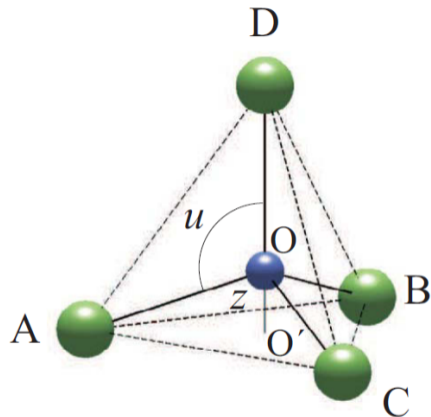
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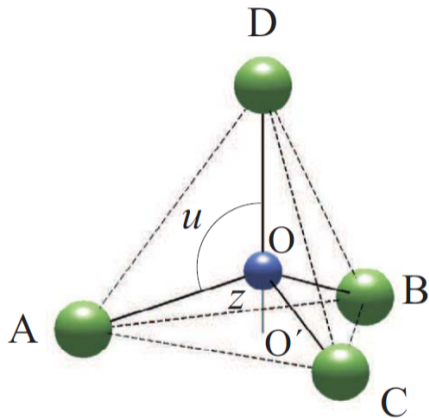
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Scattering from molecules



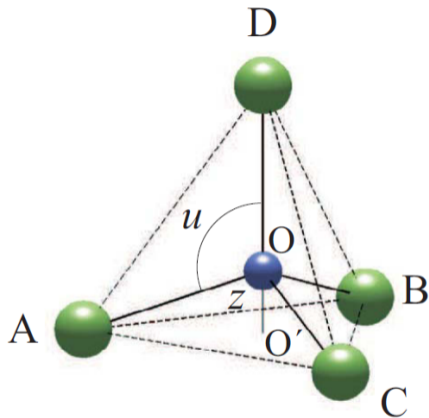
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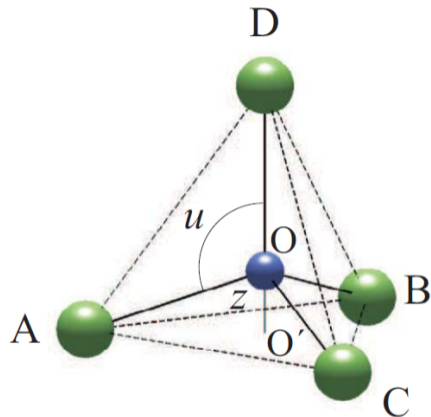
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$$\overline{OA} \cdot \overline{OD} = 1 \cdot 1 \cdot \cos u$$

Scattering from molecules



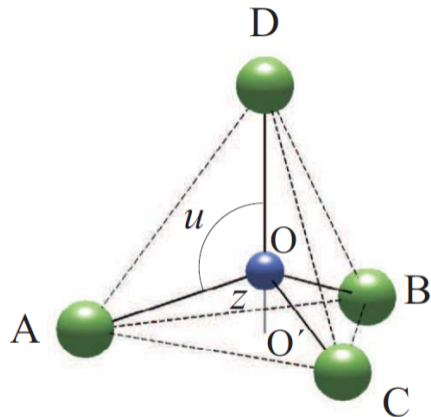
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Scattering from molecules



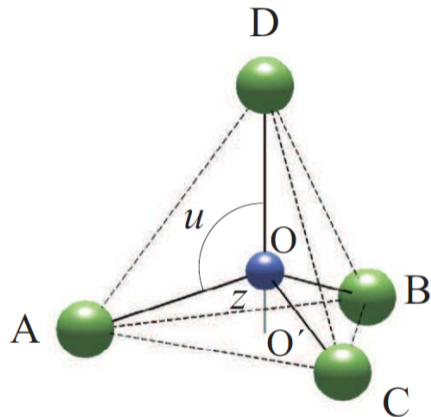
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Scattering from molecules



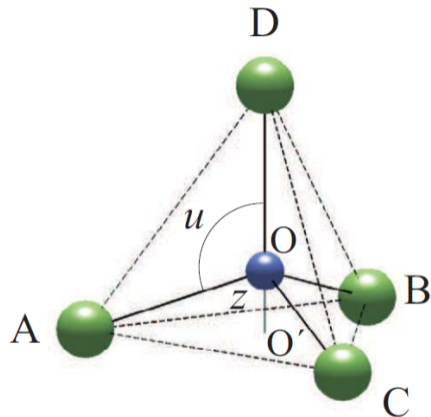
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$$\begin{aligned} |\overline{OA}| &= |\overline{OB}| = |\overline{OC}| = |\overline{OD}| = 1 \\ \overline{OA} &= \overline{OO'} + \overline{O'A} \end{aligned}$$



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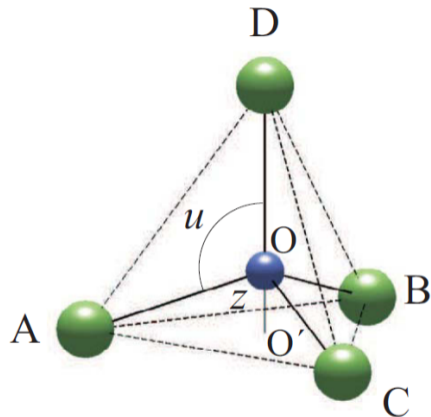
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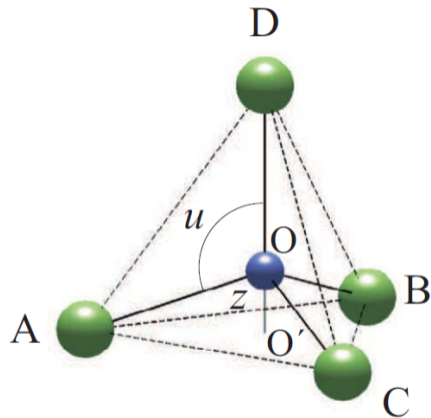


$$\begin{aligned} \overline{OA} \cdot \overline{OD} &= 1 \cdot 1 \cdot \cos u = -z \\ &= \overline{OA} \cdot \overline{OB} \end{aligned}$$

The CF_4 scattering factor



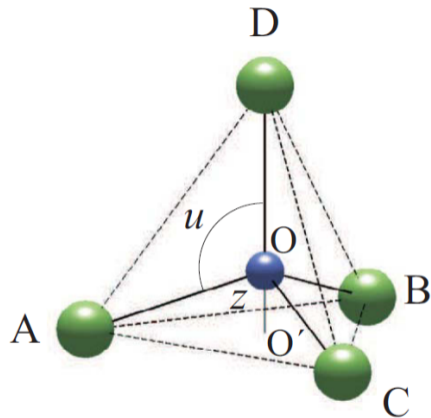
$$-z = (\overline{OO'} + \overline{O'A}) \cdot (\overline{OO'} + \overline{O'B})$$



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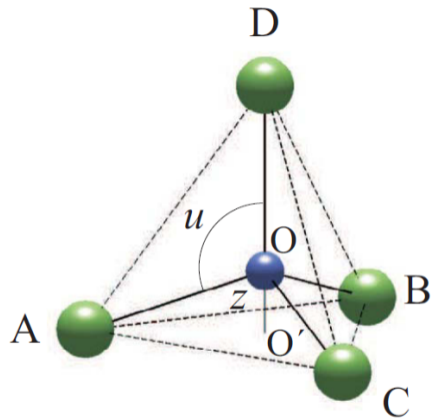
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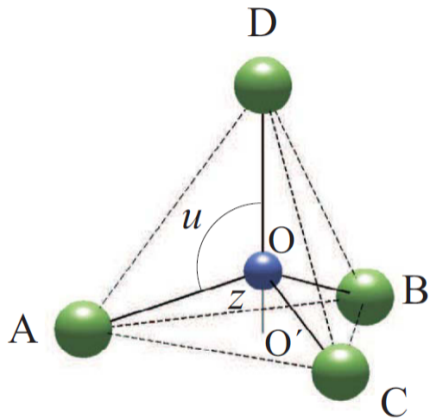
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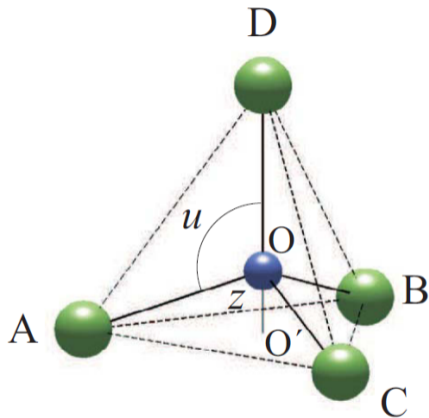


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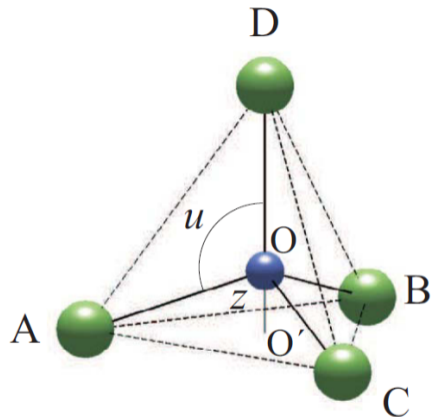
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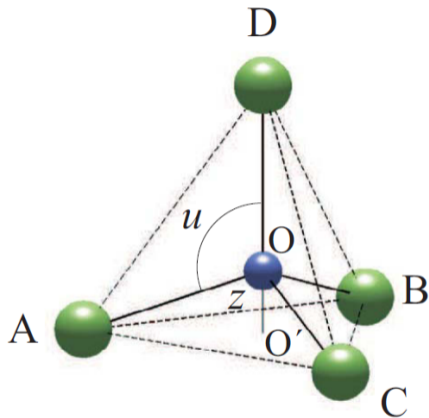
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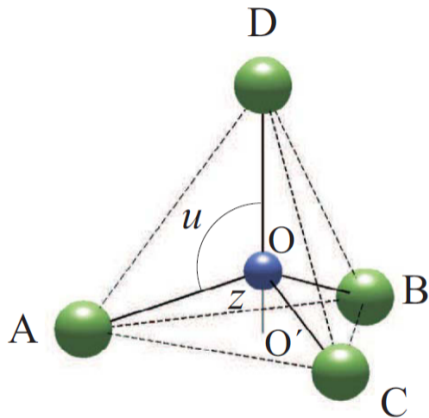
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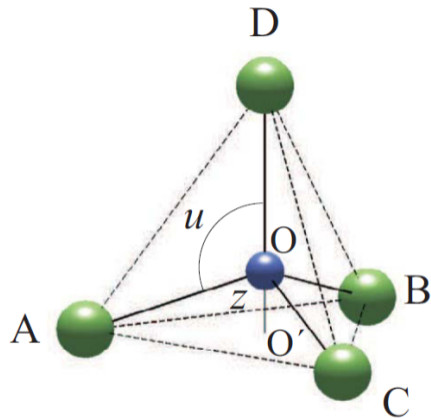
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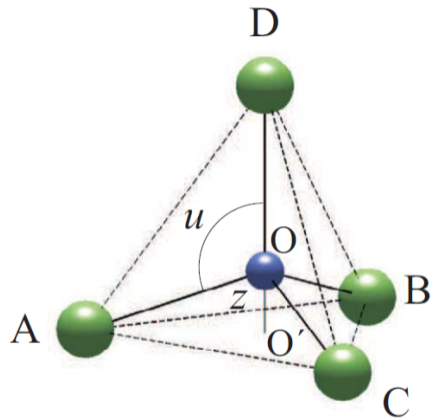


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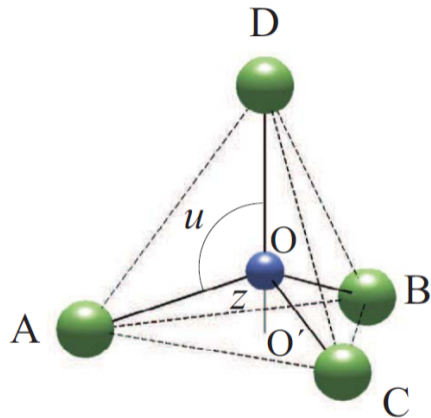
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The CF₄ scattering factor



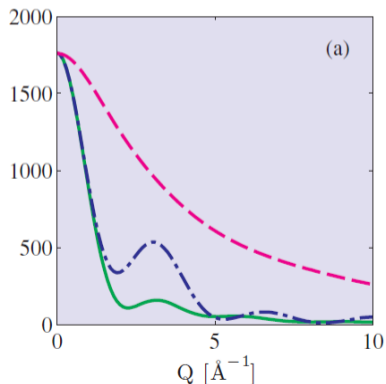
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The plot shows the structure factor of CF₄,

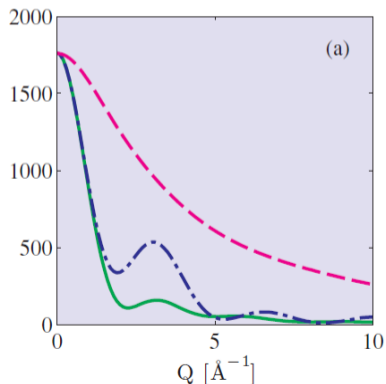


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The plot shows the structure factor of CF₄, its orientationally averaged structure factor,

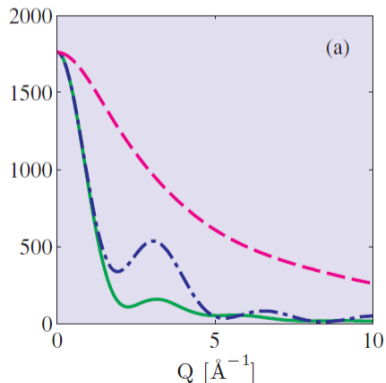
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The plot shows the **structure factor of CF₄**, its **orientationally averaged structure factor**, and **the form factor of Mo** which has the same number of electrons as CF₄

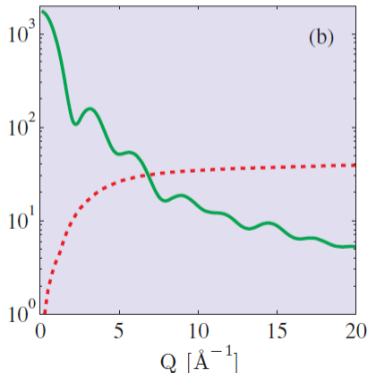
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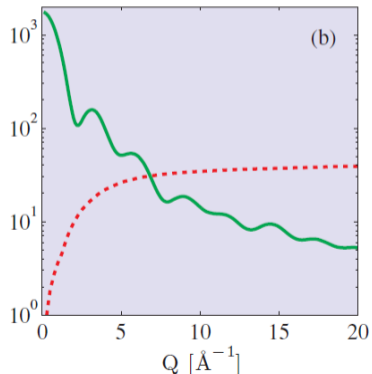
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The logarithmic plot shows the spherically averaged structure factor compared to the inelastic scattering for CF₄

The radial distribution function



The radial distribution function



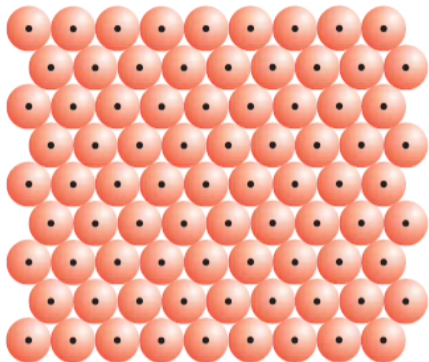
Ordered 2D crystal

Amorphous solid or liquid

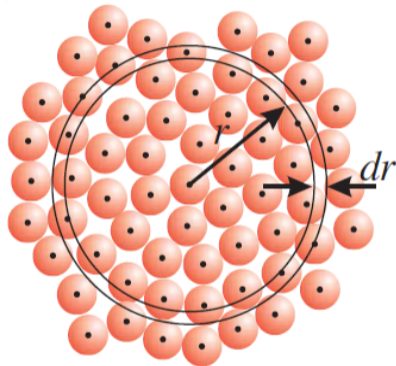
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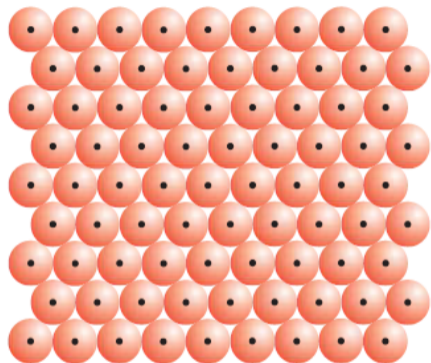
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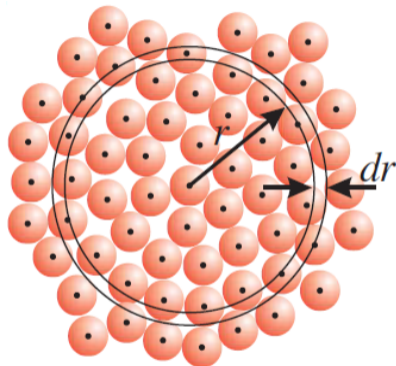
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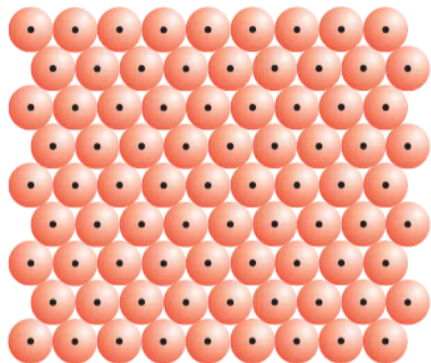


Take a circle (sphere) of radius r and thickness dr and count the number of atom centers lying within the ring.

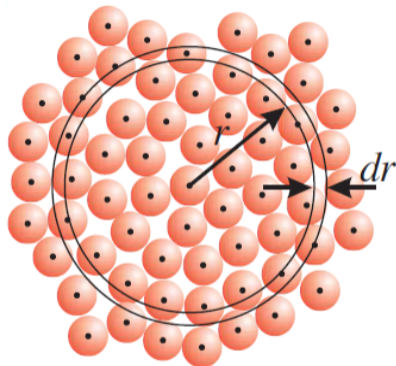
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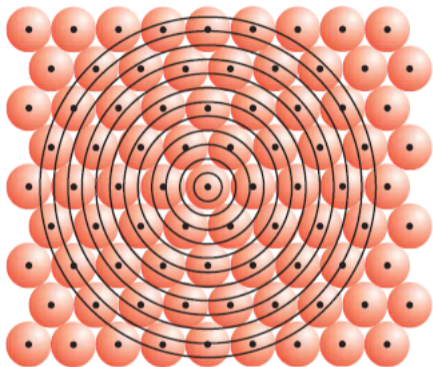


Take a circle (sphere) of radius r and thickness dr and count the number of atom centers lying within the ring. Then expand the ring radius by dr to map out the radial distribution function $g(r)$

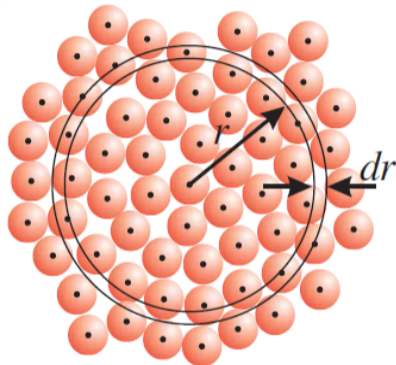
The radial distribution function



Ordered 2D crystal



Amorphous solid or liquid

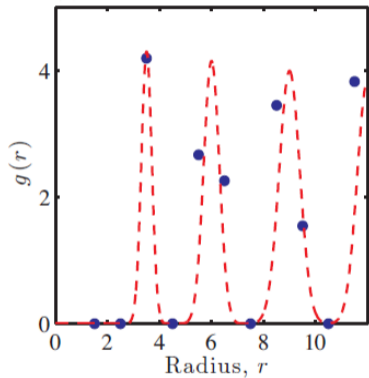


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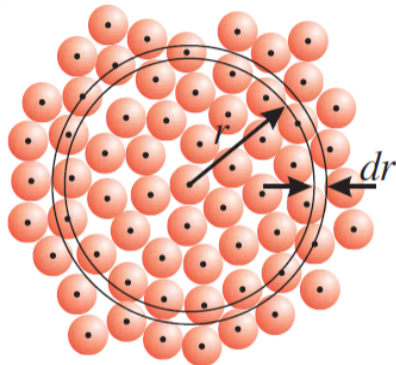
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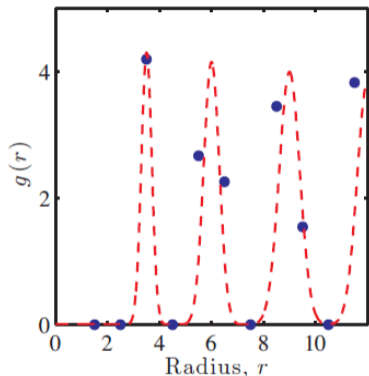


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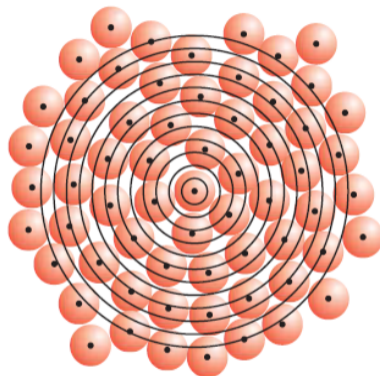
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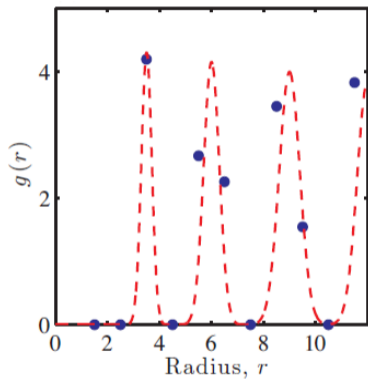


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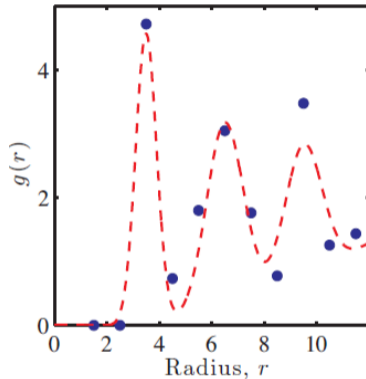
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Which is the sine Fourier Transform of the deviation of the atomic density from its average, $\mathcal{H}(r) = 4\pi r [g(r) - 1]$

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The relation between radial distribution function and structure factor can be extended to multi-component systems where $g(r) \rightarrow g_{ij}(r)$ and $S(Q) \rightarrow S_{ij}(Q)$.

Structure in supercooled liquid metals



Liquid Ni metal was suspended electrostatically and allowed to cool from its liquidus temperature of 1450°C.



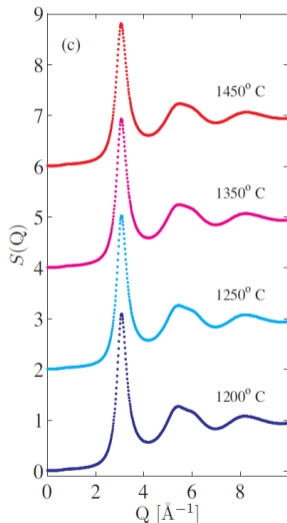
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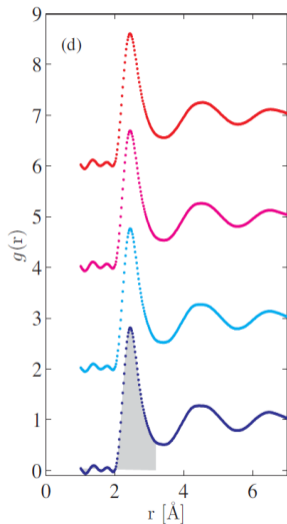
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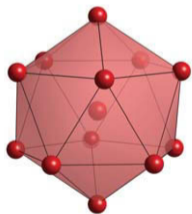
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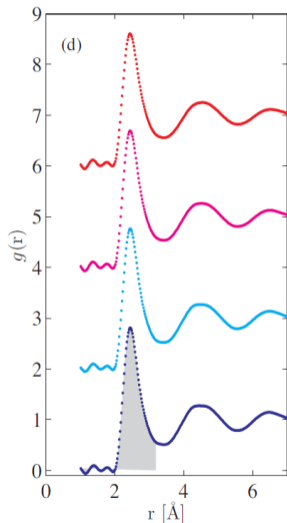
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Details in the shape of the oscillations can be indicative of distortions in the icosahedra which depend on the metal species.

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Liquid scattering can be used to study dynamics

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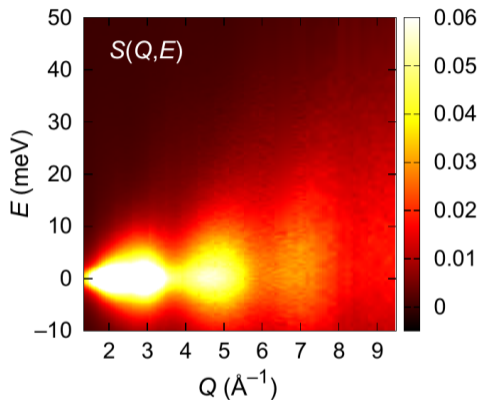
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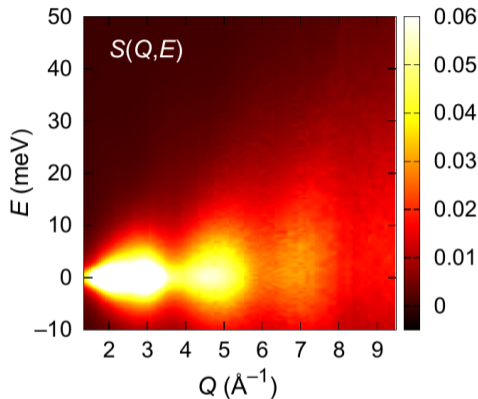
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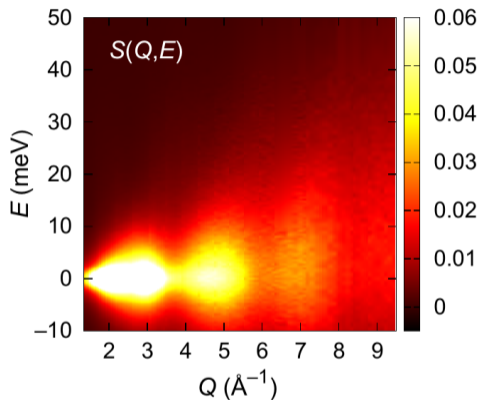


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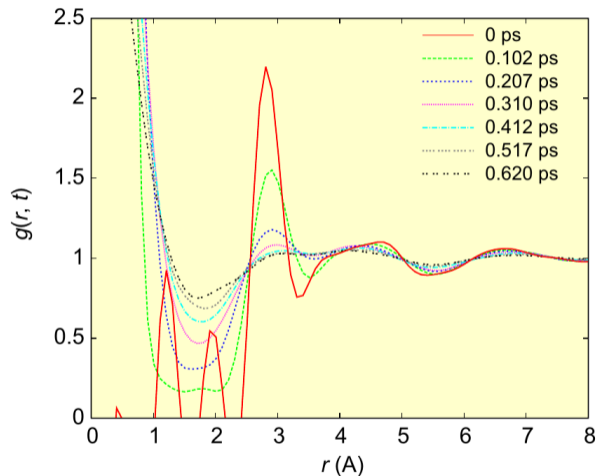
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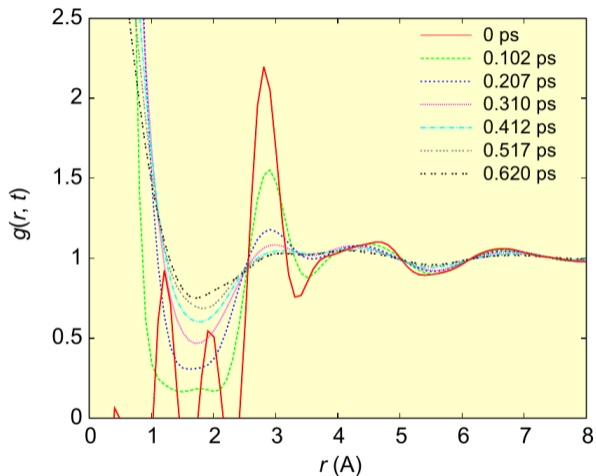


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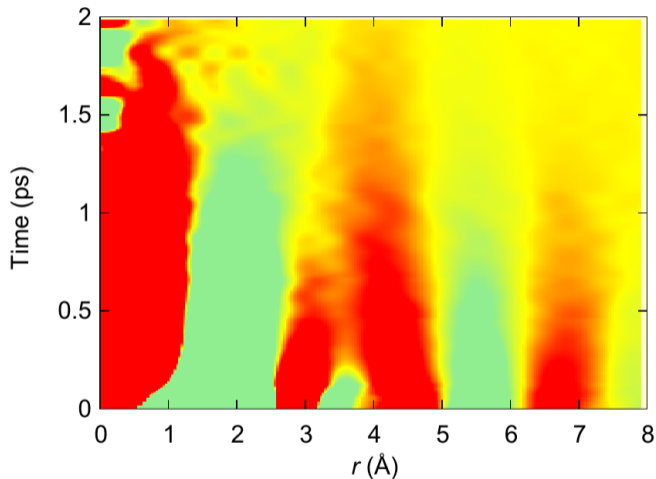


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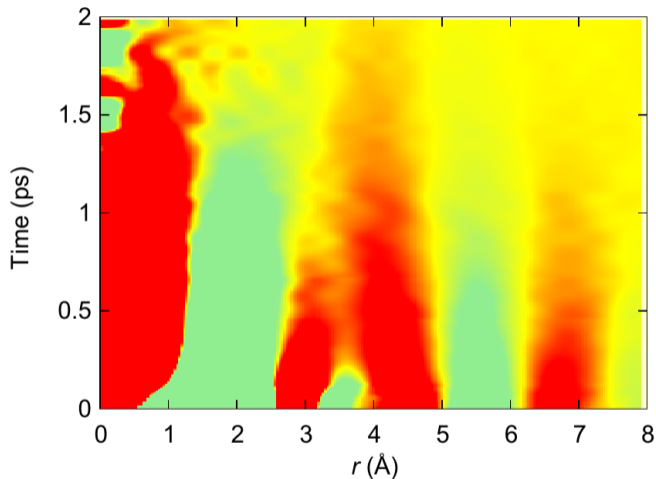
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The first and second peaks are highly coupled in space and time and merge within 0.8 ps. This behavior is different from liquid metals and leads to the viscosity of water.

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Small angle x-ray scattering



$$I(\vec{Q}) = Nf(\vec{Q})^2 + f(\vec{Q})^2 \sum_n \int_V [\rho_n(\vec{r}_{nm}) - \rho_{at}] e^{i\vec{Q} \cdot (\vec{r}_n - \vec{r}_m)} dV_m + f(\vec{Q})^2 \rho_{at} \sum_n \int_V e^{i\vec{Q} \cdot (\vec{r}_n - \vec{r}_m)} dV_m$$

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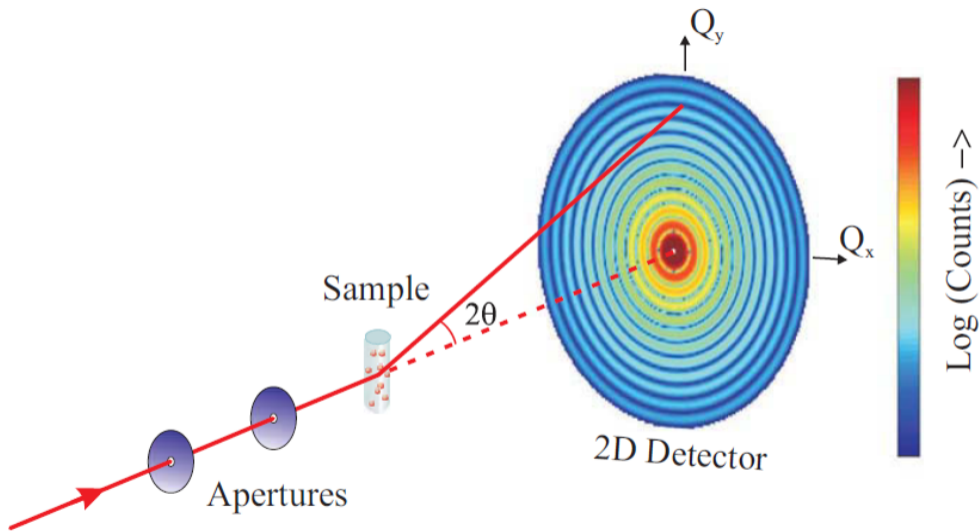
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The SAXS experiment



Scattering from a dilute solution



The simplest case is for a dilute solution of non-interacting molecules.

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Where $\Delta\rho = (\rho_{sl,p} - \rho_{sl,0})$, and the form factor depends on the morphology of the particle (size and shape).

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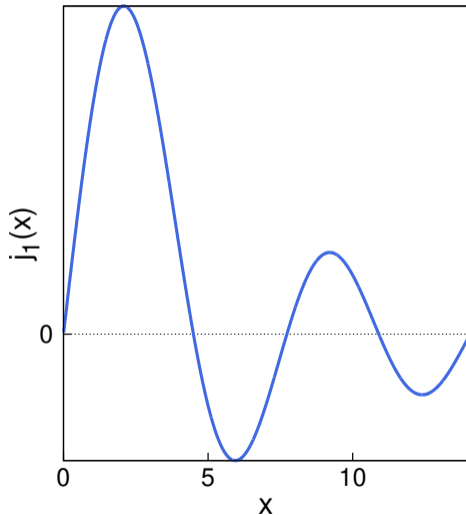


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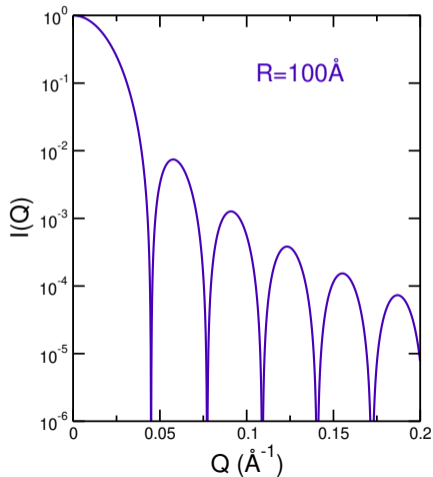


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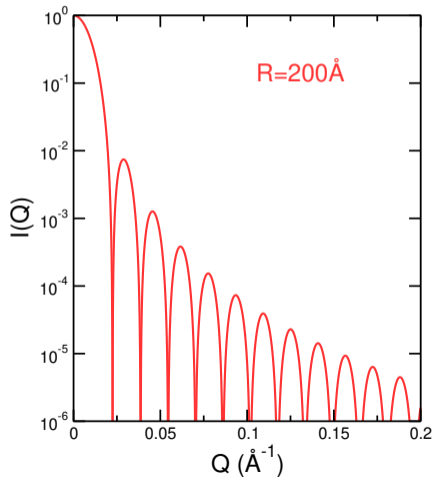
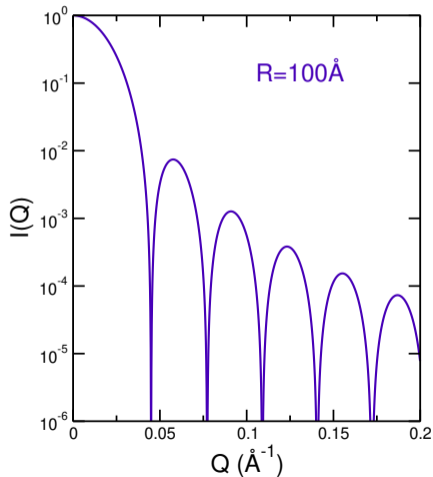
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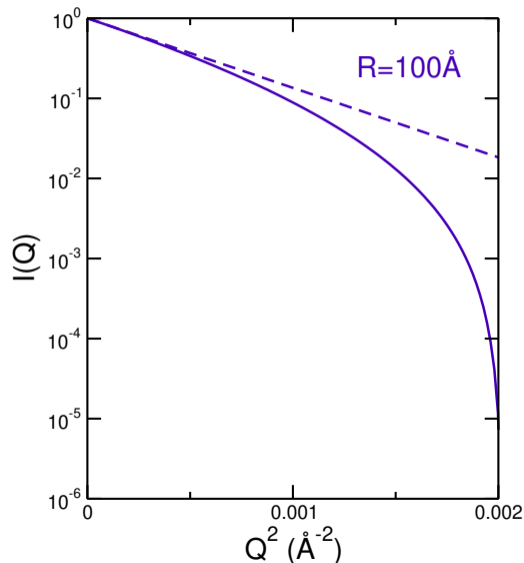


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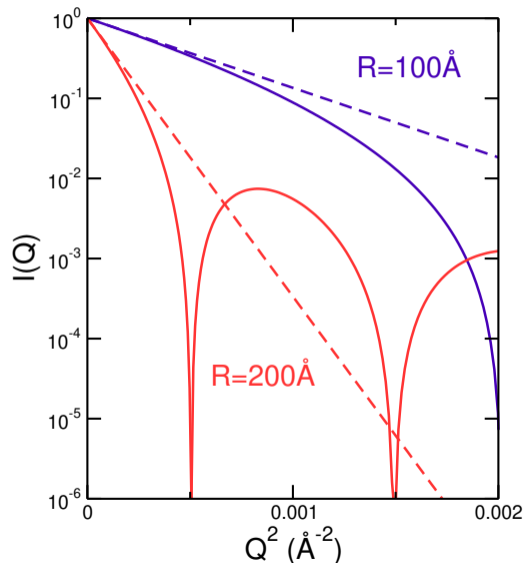


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and the initial slope of the $\log(I)$ vs Q^2 plot is $-R^2/5$



Guinier analysis

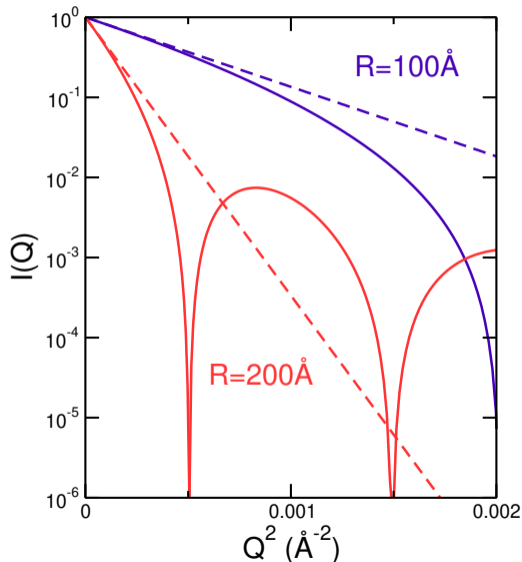


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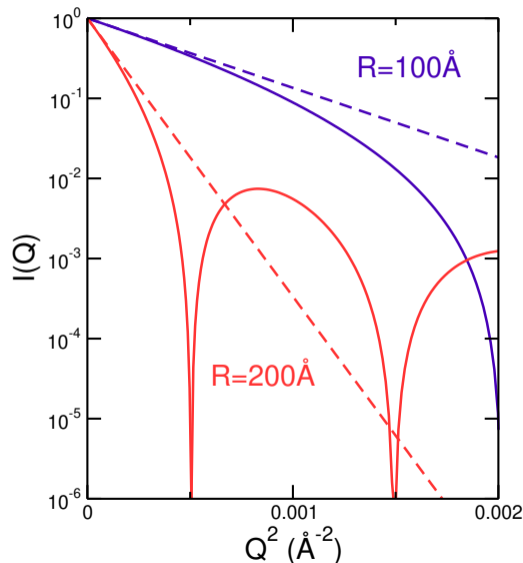
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